Total No. of printed pages = 10

3(Sem 2) CHM M2

#### 2015

### **CHEMISTRY**

(Major)

Paper : 2.2

Full Marks - 60

Time  $-2\frac{1}{2}$  hours

The figures in the margin indicate full marks for the questions.

- 1. (a) What do you mean by syn-clinal and anticlinal conformations ? 1×7=7
  - (b) Draw the most stable conformer of 1, 2difluoroethane.
  - (c) How will you prepare Gilman's Reagent for the synthesis of alkane ?
  - (d) Arrange the stabilities of conjugated diene, isolated diene and cumulated diene in decreasing order.

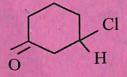
- (e) Arrange the following derivative of carboxylic acid in order of reactivity with nucleophiles. (increasing order)
   RCONH<sub>2</sub>, RCOCl, (RCO)<sub>2</sub>O, RCOOR<sup>'</sup>
- (f)  $CH_3 CH_2 CH_2 NH_2$  boils at 49°C whereas  $(CH_3)_3N$  boils at 3°C. Explain.
- (g) Nitroalkanes and nitroarenes are good solvents for polar compounds. Explain.
- 2. Answer any four questions : 2×4=8
  - (a) Draw the corresponding Newman and Saw horse projection of the following molecule.

$$HO \xrightarrow{CH_3} H$$

$$H \xrightarrow{H} Br$$

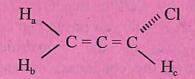
$$CH_3$$

(b) Assign R or S designation with IUPAC nomenclature of the following molecule.



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- (c) Draw the different conformations of cyclohexane and also draw the potential energy curve for the different conformers.
- (d) Identify the faces present in E and Z, but-2-ene with a suitable reaction.
- (e) What do you mean by Re and Si-face ? Give an example.
- (f) Find the topicity of the hydrogens in



# 3. Answer any *three* questions : $5 \times 3 = 15$

- (a) What is ortho-effect ? Explain on this basis why nearly all ortho substituted benzoic acids are stronger acid than benzoic acid. 2+3=5
- (b) What is ipso attack ? Explain with an example. Write the product of the following reaction 2+2+1=5

$$\frac{Br_2/light}{Or NBS}$$

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(c) Write the products in each case and give mechanism.  $1\frac{1}{2}+1\frac{1}{2}+2=5$ 

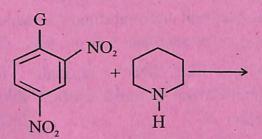
(i) Cis-2-butene 
$$\xrightarrow{BR_2}_{CCl_4}$$

(ii) trans-2-butene  $\xrightarrow{\text{OsO}_4, \text{Py}}_{\text{NaHSO}_3/\text{H}_2\text{O}}$ 

(iii) RCH<sub>2</sub>NO<sub>2</sub> + R'CHO  $\xrightarrow{\text{KOH, H}_2O}$ 

(d) How will you prepare cinnamic acid from benzaldehyde by Perkin Reaction ? Write mechanism. How will you prove that only  $\alpha$ hydrogen atoms of the anhydride are involved during the condensation reaction. 1+3+1=5

(e) In the following reaction find the products. 1+4=5



The rate of the reaction does not charge appreciably with change in the nature of G, except when G = Fluorine. Explain.

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## 4. Answer any three questions : 10×3=30

- (a) (i) What happens when an allene  $CH_2 = C=CH_2$  is treated with dil.  $H_2SO_4$  ? Give reaction. 2
  - (ii) Cyclopentadiene has an active hydrogen. To show this give a reaction. 2
  - (iii) How will you prepare an alkane by Hunsdiecker reaction ? Give probable mechanism.
  - (iv) What do you mean by transesterification reaction? Give one example.

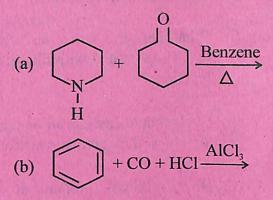
1+1=2

- (v) Carbonic acid (pKa=6) is stronger acid than phenol (pKa=9.95). Explain. 1
- (b) (i) In Rosenmund reduction of RCOCl to RCHO the catalyst  $Pd/BaSO_4$ ,  $H_2$  is used along with small amount of sulphur or quinoline. Explain. 2
  - (ii) What happens when propionaldehyde is treated with Al(OEt)<sub>3</sub>? What is the name of the reaction ?

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(5)

(iii) Complete the following reactions: 2



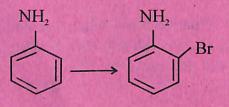
 (iv) Arrange the following group in increasing activating order towards electrophilic reagent with explanation.

<sup>®</sup>NR<sub>3</sub> NH<sub>2</sub>, NHCOCH<sub>3</sub>

 (v) Formaldehyde undergoes Cannizzaro reaction, but acetaldehyde does not. Explain.

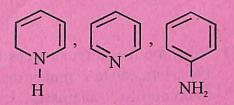
2

- (c) (i) How will you protect a NH<sub>2</sub> group in aniline during nitration ? 1
  - (ii) Convert the following:



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(iii) Arrange the following in increasig order of basicity : 1



(iv) Write the products with mechanism and name the following reactions:  $2 \times 3=6$ 

(i) Ph CHO + Br CH<sub>2</sub>CO<sub>2</sub> $\varepsilon t \xrightarrow{Zn} H_3O^+$ 

(ii) 
$$CH_3 \rightarrow O - CHO + CH_2(CO_2 \varepsilon t)_2$$
  
$$\frac{Me_2 NH_2}{MeCOO^-}$$

(d) (i) What do you mean by Kinetic Isotope effect ? How this effect can be applied as evidence for Areniumion mechanism in ArSE, reaction ? 1<sup>1</sup>/<sub>2</sub>+1<sup>1</sup>/<sub>2</sub>=3

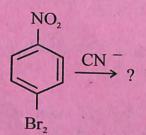
(ii) Benzene does not decolorise bromine
 water. Explain.
 2

(iii) Friedel Crafts acylation of aniline is difficult. Briefly explain. 2

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 (iv) How will you explain that electrophilic substitution in anthracene takes place almost exclusively at 9 or 10 position.
 3

- (e) (i) What are the different steps involved in the intermediate complex mechanism of nucleophilic aromatic substitution reaction? What are the evidences in support of this mechanism ? 2+2=4
  - (ii) Write down the mechanism of ArSN reaction involving benzyne intermediate.
  - (iii) How will you trap a benzyne intermediate? 2
  - (iv) Write the product of the following reaction : 1



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(f) Identify the products in the following reactions : 6

(i)

$$\begin{array}{c} \text{CHO} \\ \hline \\ \hline \\ + (\text{CH}_3)_2 \text{ C} = \text{O} \xrightarrow{\text{KNH}_2} \text{A} \xrightarrow{\text{NaBH}_4} \text{B} \end{array}$$

 $\overset{\text{NO}_2}{\Gamma} + \overset{\text{Sn}}{\xrightarrow{\text{ConcHCl}}} C \xrightarrow{\text{n-pentyl}} D \xrightarrow{\text{H}_3\text{PO}_2} E$ 

$$\bigcirc \stackrel{\text{OH}}{\longleftarrow} + H_2O \longleftrightarrow F \xrightarrow{\text{FeCl}} G$$

$$CH_{3}CH = CH_{2} \xrightarrow{H^{+}, CO} H \xrightarrow{RNH_{2}} I$$

$$\bigcirc \bigcirc \bigcirc \xrightarrow{\text{CH}_3\text{COCl}} J \xrightarrow{\text{Zn(Hg)}} K$$

$$O \xrightarrow{CH_3} PCC \xrightarrow{PCC} L$$

 (ii) Benzene is not used as a solvent for the Friedel Crafts alkylation of chlorobenzene. Explain.

(iii) Between phenol and benzylalcohol which one is stronger acid ? Give reason. 2