2015

CHEMISTRY

(Major)

Paper: 5.3

Full Marks: 60

Time: 3 hours

The figures in the margin indicate full marks for the questions

1. Answer the following questions (any seven):

 $1 \times 7 = 7$

(a) Give the product of the following reaction:

CH₃—CH=CH—COOH
$$\frac{1) \text{ LiAlH}_4, \text{ Ether}}{2) \text{ H}_3\text{O}^+}$$

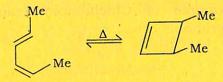
- (b) Which among furan, pyrrole and thiopene undergoes Diels-Alder reaction? Write the structure of the product formed by it with maleic anhydride.
- (c) Why does electrophilic aromatic substitution of indole occur preferably at the 3-position?

- (d) Which bond of phenanthrene is readily attacked by reagents?
- (e) 2-methyl-2-nitropropane does not dissolve in alkali whereas 2-nitropropane dissolves. Why?
- (f) Write the structure of the product when the following compound is treated with Na/Hg:

PhCH=CHCH₂N(CH₃)₃P

- (g) Why does pyridine not undergo Friedel-Crafts reaction?
- (h) Why is catalytic reduction of thiophene difficult?
- 2. Answer the following questions (any four): 2×4=8
 - (a) Write the appropriate product for the following reactions:

- (b) Draw the tautomers of acetoacetic ester. Identify the stable form and explain why it is more stable than the other form.
- (c) How will you use 1,3-dipolar reagents to synthesize five-membered heterocyclic compounds?
- (d) Which one is more acidic—ethanethiol or ethanol and why? How can one distinguish ethanethiol from ethanol?
- (e) Why is cope rearrangement called [3,3] sigmatropic shift?
- 3. Answer any three of the following questions [any one from (a) and (b), any two from (c), (d) and (e)]: 5×3=15
 - (a) How can you convert propanoic acid to ethanamine, using a reaction that involves isocyanate intermediate? Name the reaction. Write the mechanism of the reaction. 2+1+2=5
 - (b) Explain why (2E, 4Z) hexadiene thermally cyclizes to give cis-3,4-dimethyl cyclobutene.



Write the product of the following reaction:

4+1=5

(c) How will you distinguish between 1°, 2° and 3°-nitroalkanes? What products are obtained when nitrobenzene is reduced in (i) acidic and (ii) alkaline media?

3+2=5

- (d) Explain why methylene group in diethylmalonate is more reactive than methylene group in malonic acid. Starting from diethylmalonate, how can you synthesize (i) a dicarboxylic acid, (ii) a heterocyclic compound, (iii) an alicyclic compound and (iv) α, β-unsaturated acid? 1+1+1+1=5
- (e) What are the IUPAC names of pyrrole, furan and pyridine? Write down the steps involved in the Bischler-Napieralski reaction leading to synthesis of isoquinoline. Give an example of Chichibabin reaction of pyridine.

4. Answer the following questions:

Either

(a) (i) Predict the product in each case and write the mechanism for each:

(1) CH_3 CH_3 CH_3 CH_4 CH_5 CH_5 CH_6 CH_7 CH_7 CH_8 $CH_$

(2)
$$CH_3 CH_3 \xrightarrow{H_2SO_4}$$
?

- (ii) (1) Explain why reaction of naphthalene with conc. H₂SO₄ at 40 °C yields naphthalene-1-sulphonic acid whereas at 160 °C naphthalene-2-sulphonic acid is the major product.
 - (2) Convert nitrobenzene to 1,3-dichlorobenzene (give equations).

Or

(b) (i) Give the product of the following reaction, name the rearrangement and propose a mechanism: 1+1+3=5

$$\begin{array}{c|c}
O & O \\
\parallel & \parallel \\
PhC-CPh & KOH/EtOH
\end{array}$$
?

Diphenylethanedione

3

2

(ii) Show by symmetry correlation diagram approach that [2+2] cycloaddition is a photochemically allowed process.

Either

(c) (i) Complete the following reactions:

(1)
$$O-CH_2-CH=CH_2$$
 CH_3
 $O-CH_2-CH=CH_2$
 $O-CH_2-CH=CH_2$
 $O-CH_2-CH=CH_3$
 $O-CH_3$
 $O-C$

Account for the product obtained in each case. $2\frac{1}{2}+2\frac{1}{2}=5$

(ii) Identify A, B, C, D and E in the following reactions: $1 \times 5 = 5$

$$(1) \quad \bigvee_{\text{Furan}} + \left\langle \bigcirc \right\rangle \xrightarrow{\mathbb{N}} \stackrel{\oplus}{\text{NCl}} \xrightarrow{\text{OH}} A$$

$$(2) \qquad \xrightarrow{\text{KMnO}_4} B$$

(3)
$$CH_3NO_2 \xrightarrow{Cl_2/NaOH} C$$

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(4)
$$CH_3-N=C=O+\longrightarrow D$$
(5) $CrO_3 \longrightarrow E$

Or

(d) (i) What are the disadvantages of heterogeneous catalytic hydrogenation? Predict the product in each of the following reactions:

2+1+1+1=5

(1) PhCH=CHCO₂CH₂Ph
$$\xrightarrow{[Ph_3P]_3RhCl}$$
?

(2) PhCH₂OH
$$\xrightarrow{\text{H}_2\text{-Pd}}$$
?

(3)
$$\xrightarrow{\text{H}_2; (Ph_3P)_3RhCl} ?$$

(ii) Describe the mechanism involved in the oxidation of 1,2-diols with lead tetraacetate. Identify the product in the following reactions:

(1)
$$CH_3COCH_3 + SeO_2 \longrightarrow ?$$

Either

- (e) (i) How many monosubstituted derivatives of naphthalene are possible? Which position is preferentially attacked in electrophilic substitution reactions of naphthalene? How can the following naphthalene derivatives be prepared? 1+1+3=5
 - (1) 2-naphthylamine
 - (2) 1-naphthol
 - (3) 1,4-naphthaquinone
 - (ii) Why is EAA called an active methylene compound? Starting with EAA, how can you prepare the following?

(No mechanism needed)

(Continued)

(f)	(i)	Describe a method of synthesis of furan.	3
	(ii)	How can you prepare the following?	2
		(1) by Wittig reaction	
		(2) PhCOOMe by Baeyer-Villiger reaction	
	(iii)	Why does Hofmann elimination of a quaternary ammonium salt give thermodynamically less stable	
		alkene as the predominant	

(iv) Write one general method of synthesis of thiols, RSH. How can RSH be converted to (1) RSSR and (2) RSR? 1+1+1=3

product?

2